

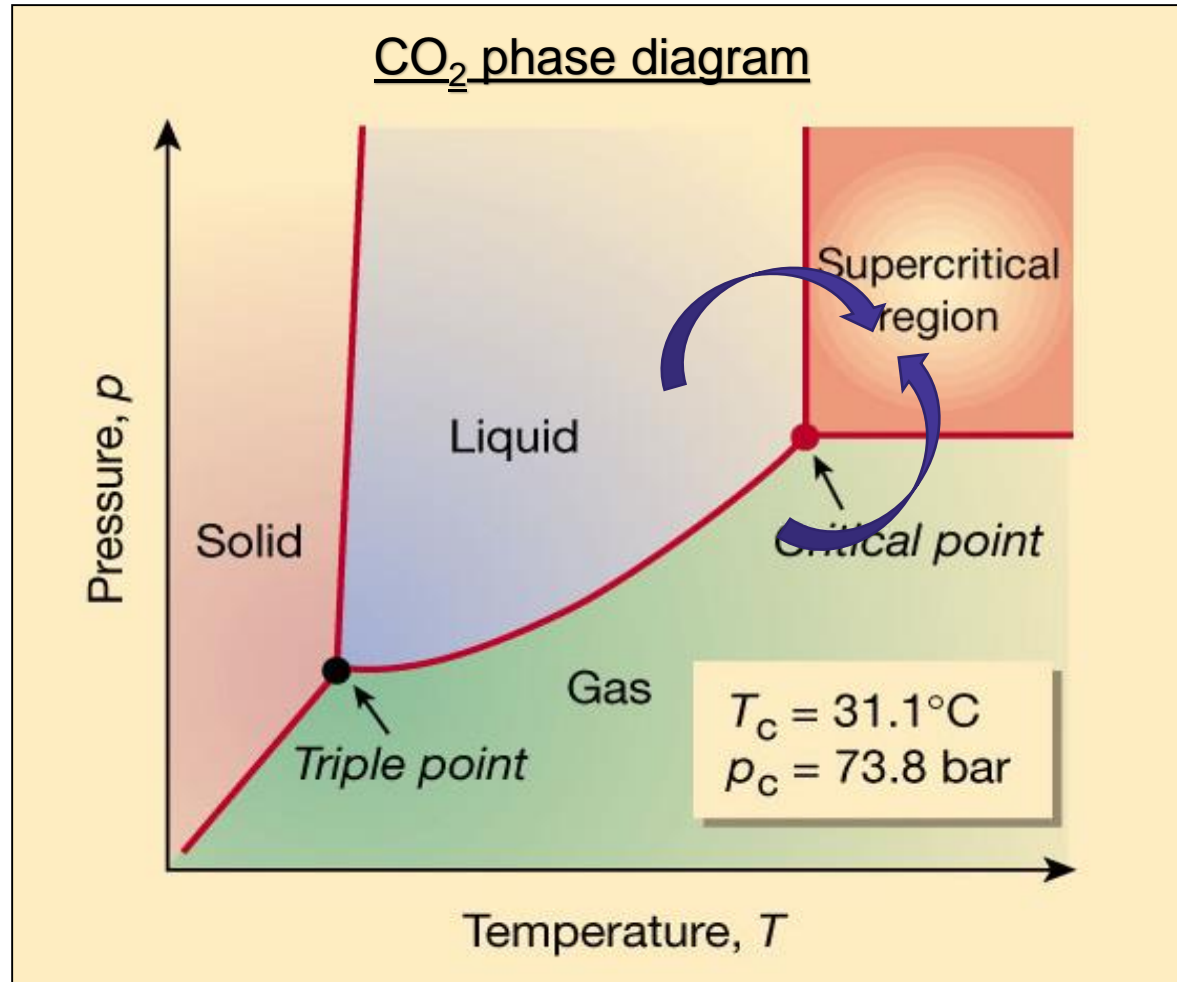
# Supercritical carbon dioxide (scCO<sub>2</sub>) processing of Ceramic Materials

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*Advanced Ceramic Materials Group*

# Contents

- Introduction to  $\text{scCO}_2$
- Potential and applications of  $\text{scCO}_2$  processing
- Research in ceramic materials group

# What is supercritical CO<sub>2</sub>



Solid



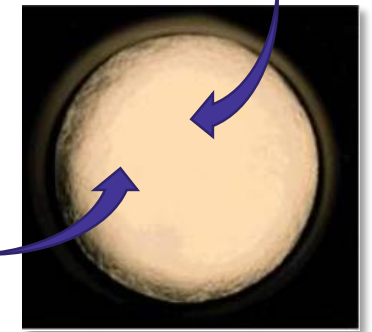
Liquid



Gas



Supercritical



# Supercritical CO<sub>2</sub> (sccO<sub>2</sub>)



Gas-like

*Properties from both*

Liquid-like

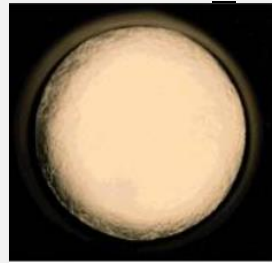
- High diffusivity & mass transfer
- Flowing properties (e.g. viscosity)
- No surface tension

- High density



Flow properties &  
Diffusion through structures

scCO<sub>2</sub>



Better solvability &  
Tunable solvability

Other processing benefits:

Inexpensive

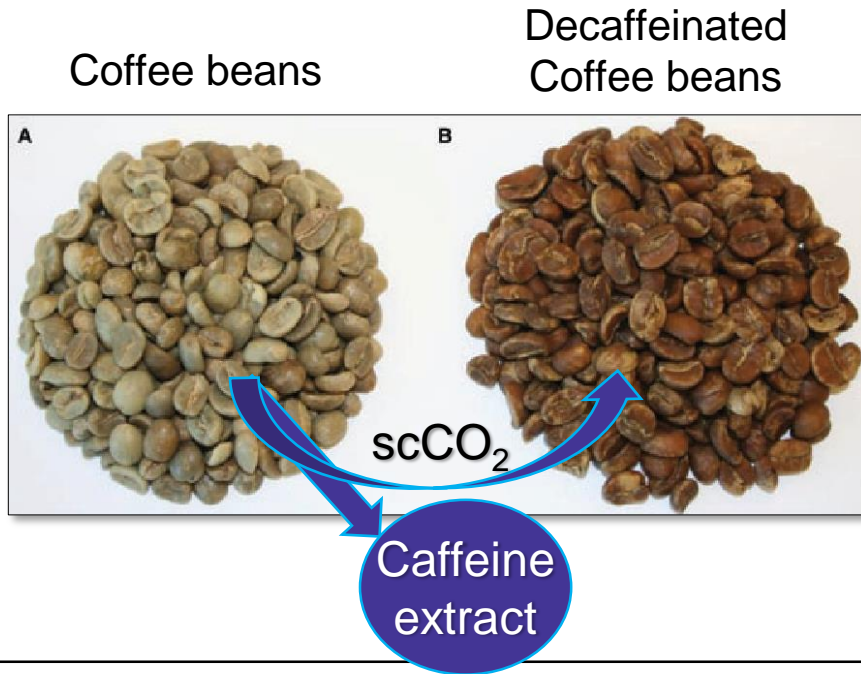
Easily  
available

Low  
supercritical  
point

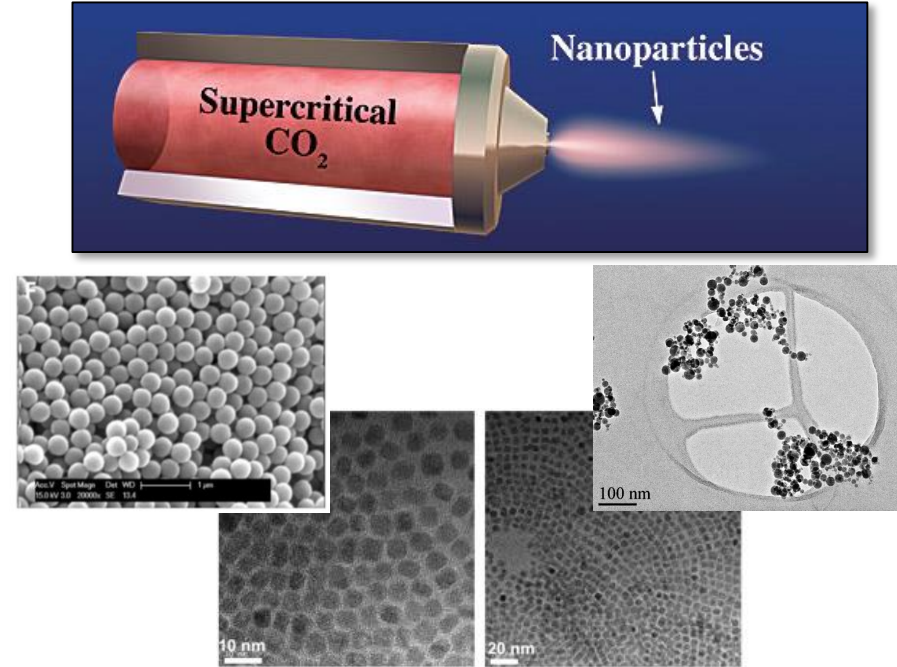
Non-toxic

# scCO<sub>2</sub> application areas

## Extraction



## Nanomaterial synthesis and preparation



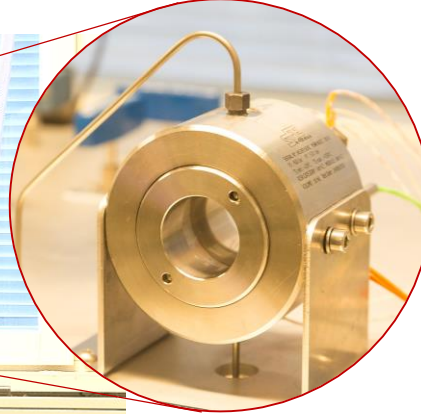
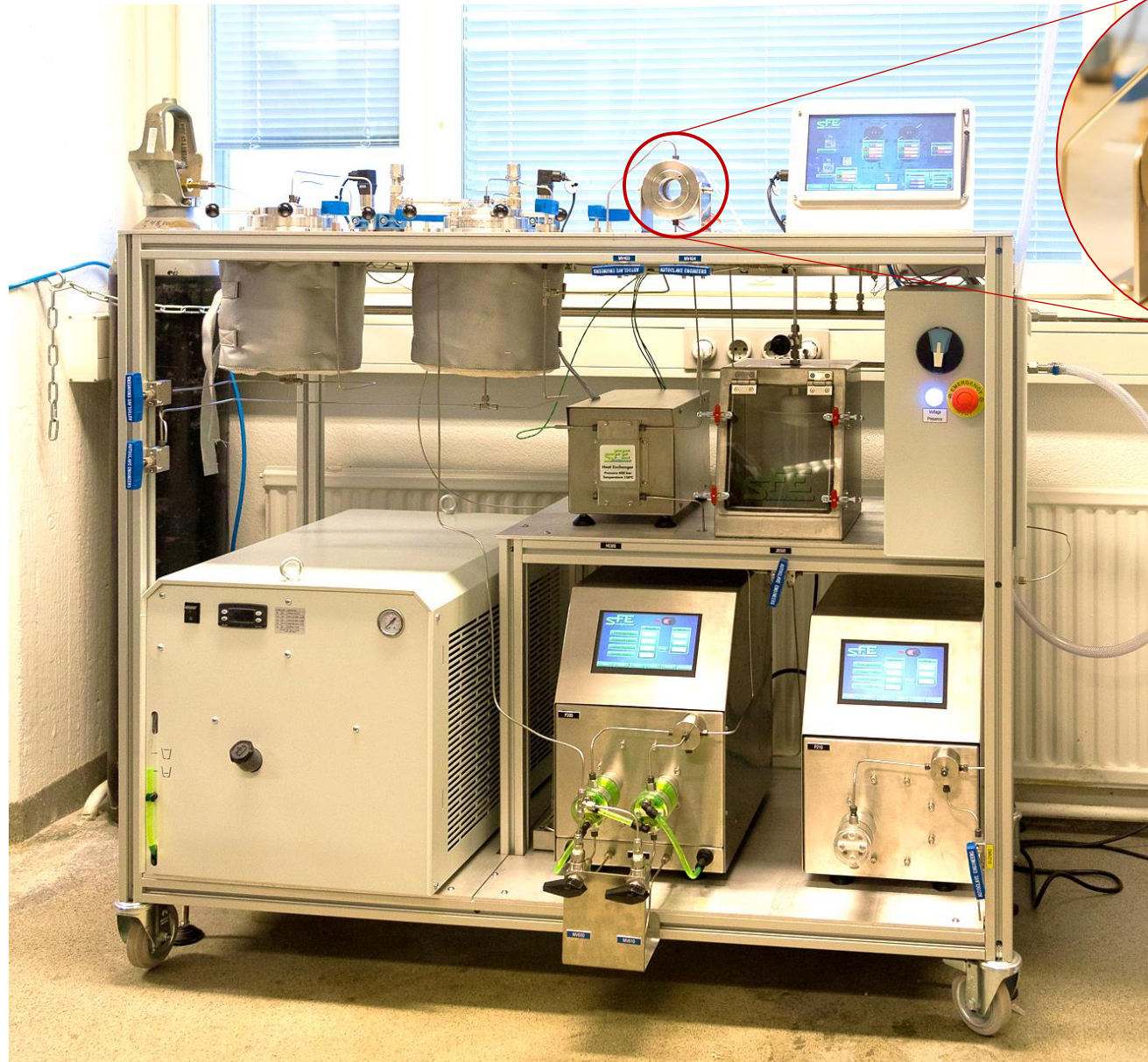
Other areas:

- Polymer/chemical synthesis
- Carbon storage
- Pharmaceuticals
- Dry cleaning
- Etc...

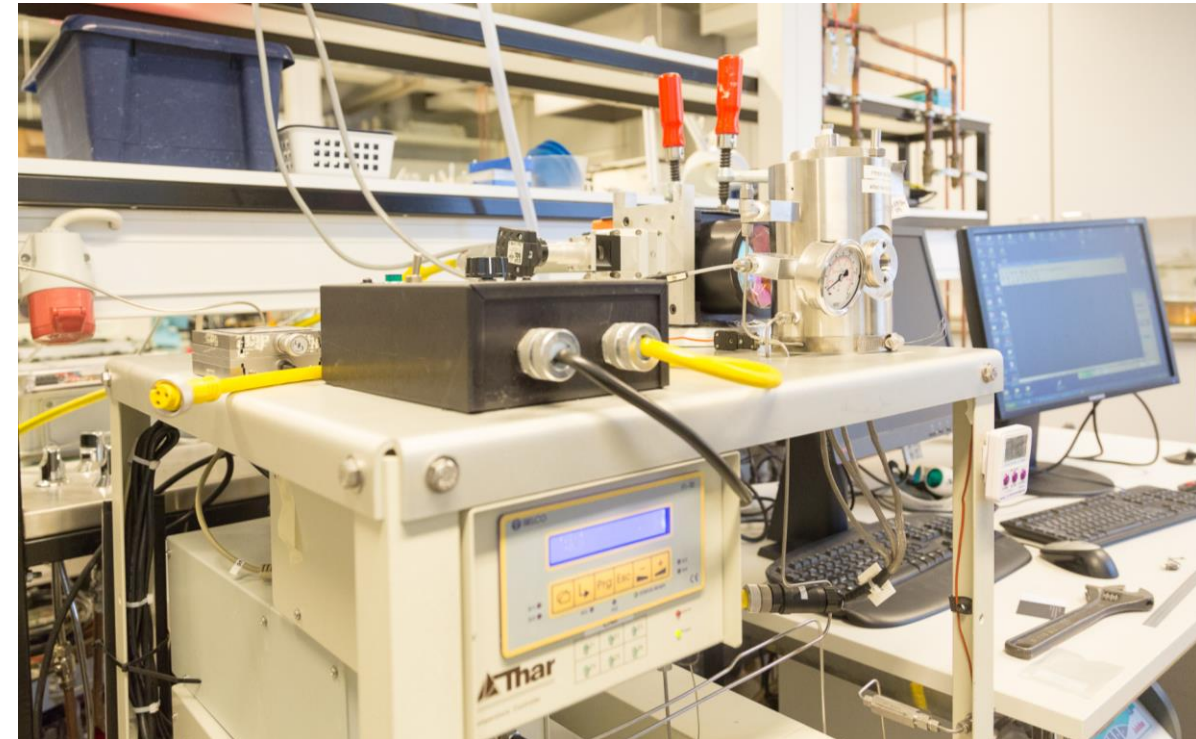
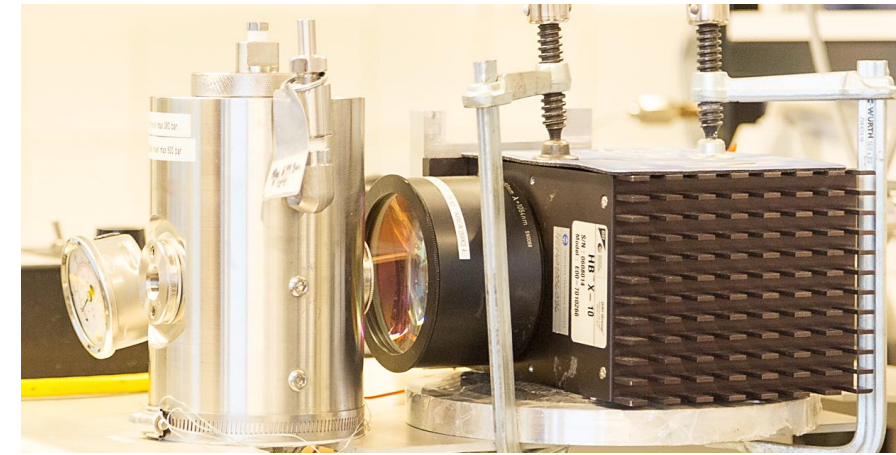


# Our equipment

## System 1



## System 2



# scCO<sub>2</sub> research in Ceramic Group

**1. SiO<sub>2</sub> Aerogel drying**

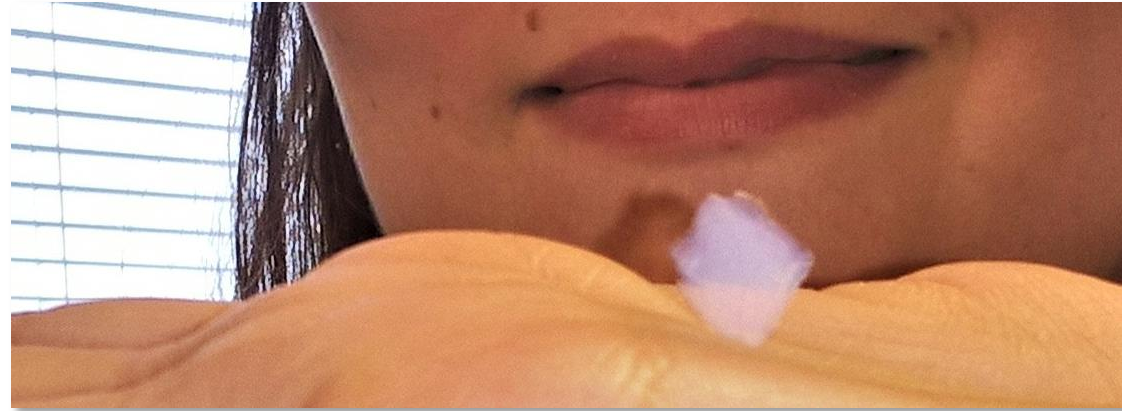
**2. Artificial zinc patina**

- a) Pretreatment for painting
- b) Conversion to ZnO

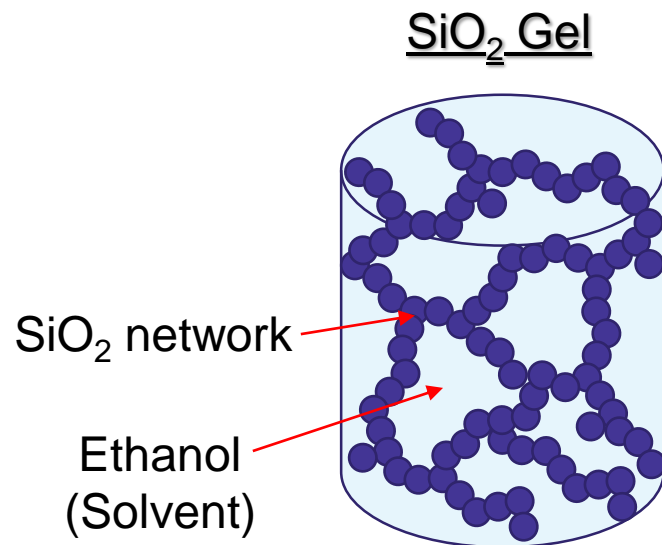
**3. Laser ablation of Titanium target**



# 1. SiO<sub>2</sub> aerogel drying



© Sandiegouniontribune, 2010, Mae Anderson



Solvent removal  
with sCCO<sub>2</sub>

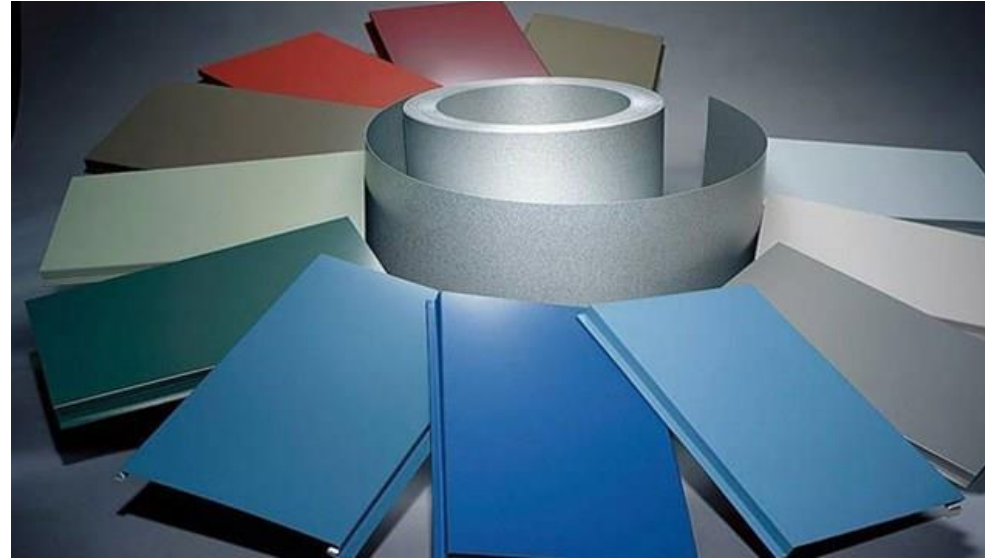


Aerogel

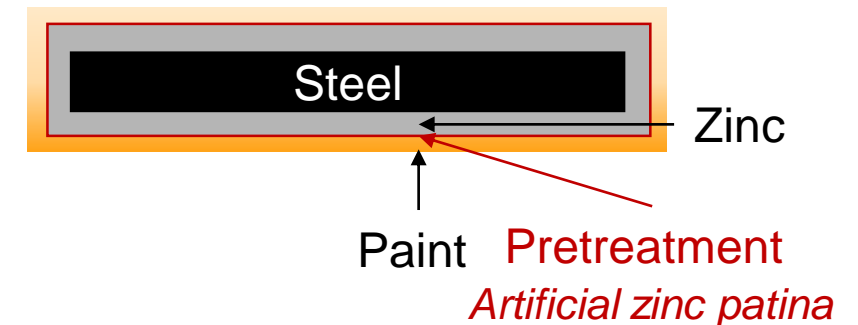
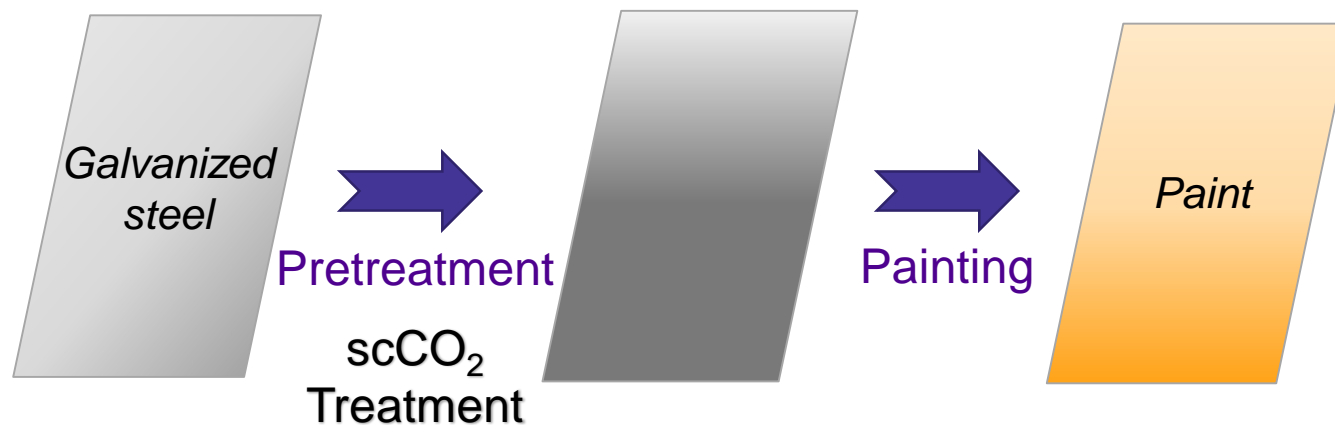
- No aging
- Porosity ~ **97 v-%**
- Surface area **598 m<sup>2</sup>/g**



## 2. Artificial zinc patina: a) *Pretreatment for painting*



### Painting process



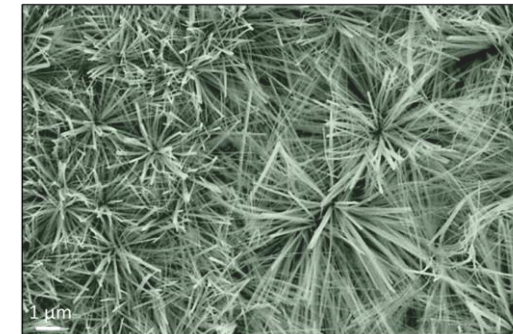
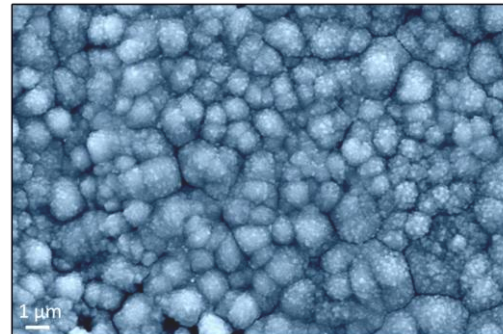
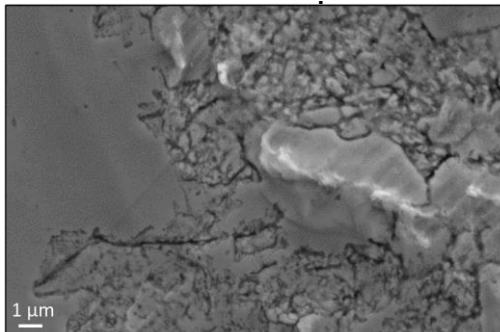
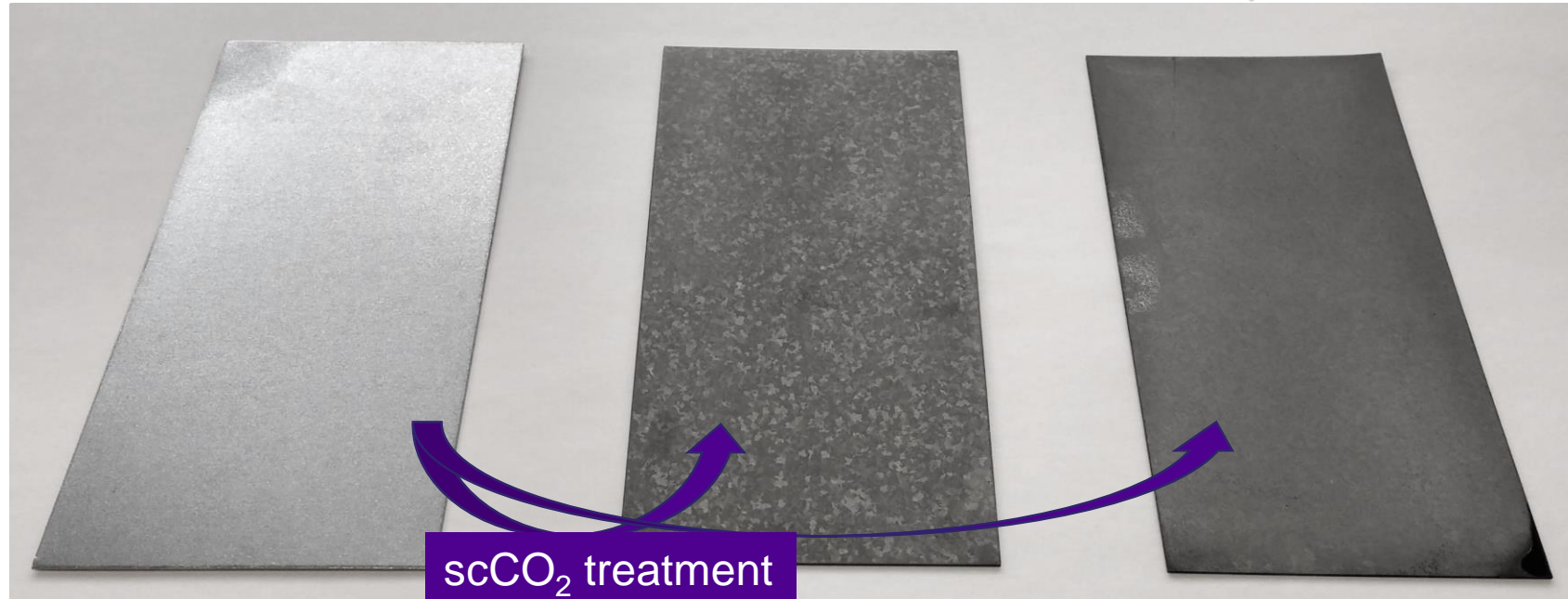
## 2. Artificial zinc patina:

### a) *Pretreatment for painting*

*Galvanized steel*

*Dense patina*

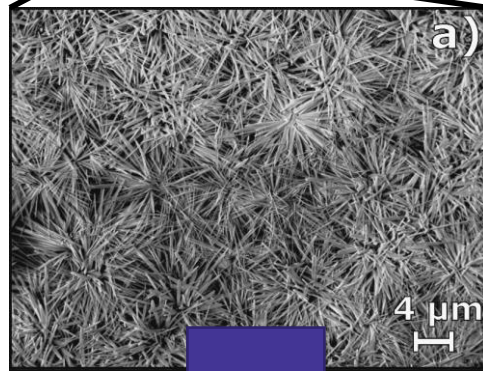
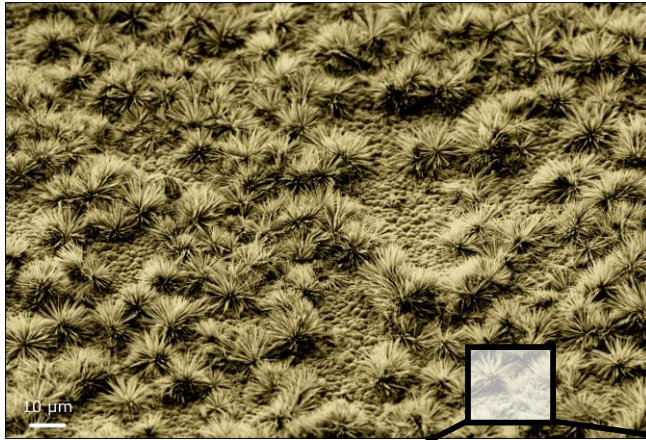
*Porous patina*



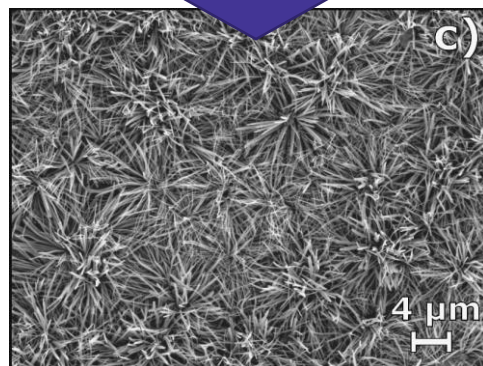


## 2. Artificial zinc patina

### *b) Conversion to ZnO*



300°C

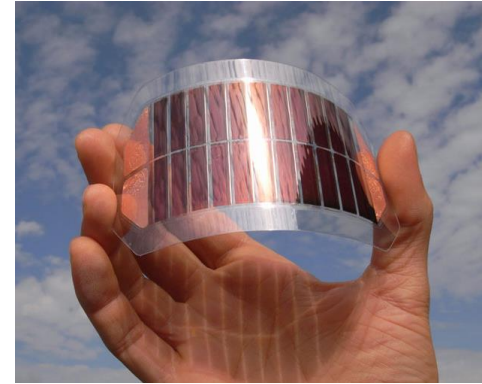


**Zinc patina**  
( $\text{Zn}_x(\text{CO}_3)_y(\text{OH})_z$ )

Release of  
 $\text{CO}_2$  and  $\text{H}_2\text{O}$

**Zinc oxide**  
( $\text{ZnO}$ )

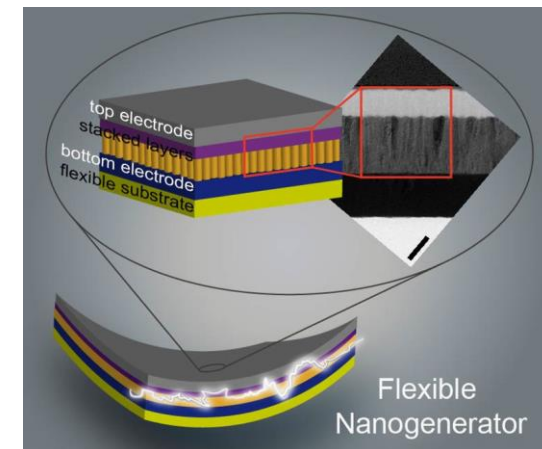
DS Solar cells



Gas sensor



Energy harvesting



Applications

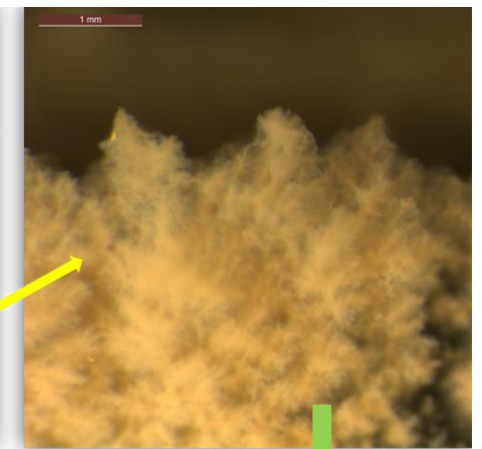
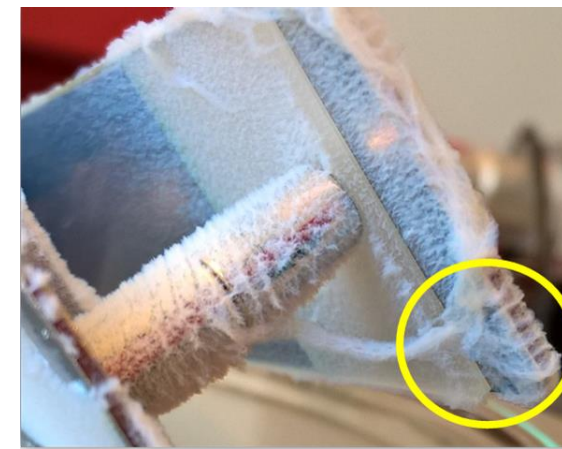
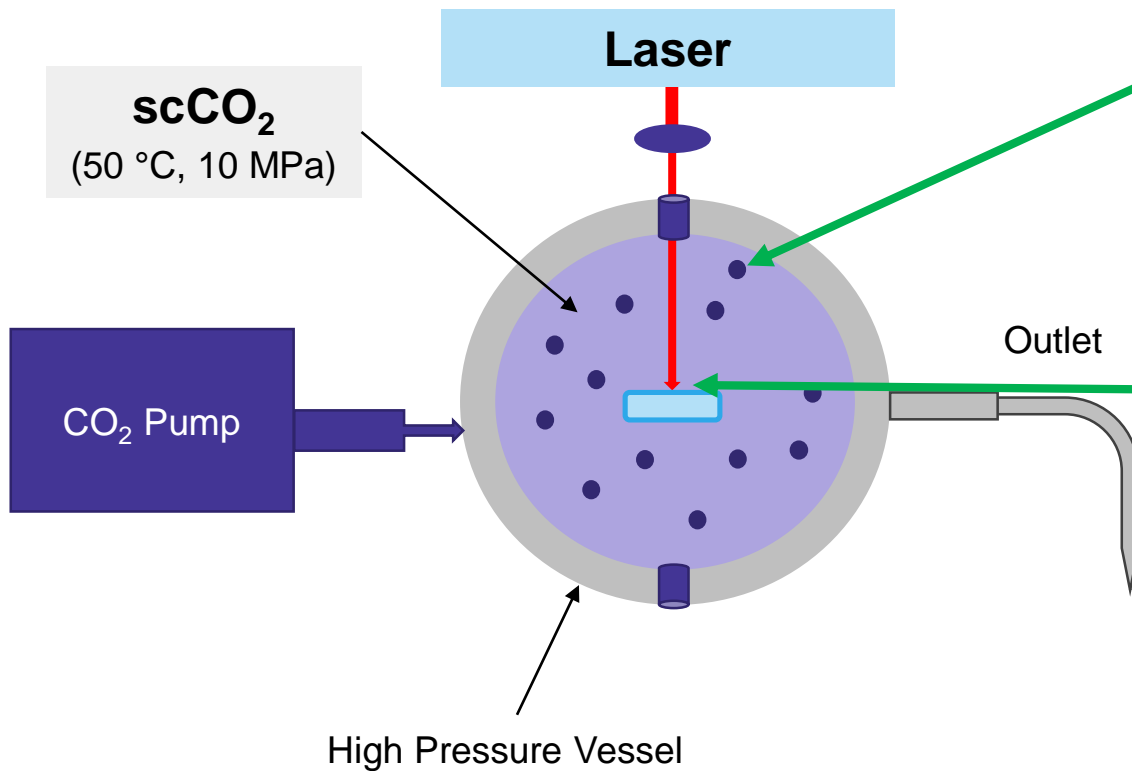


# 3. Pulsed laser ablation in scCO<sub>2</sub>

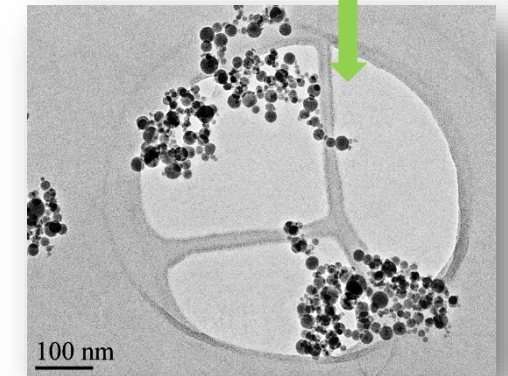
Photoactive nanoparticles &  
Functionalized surfaces



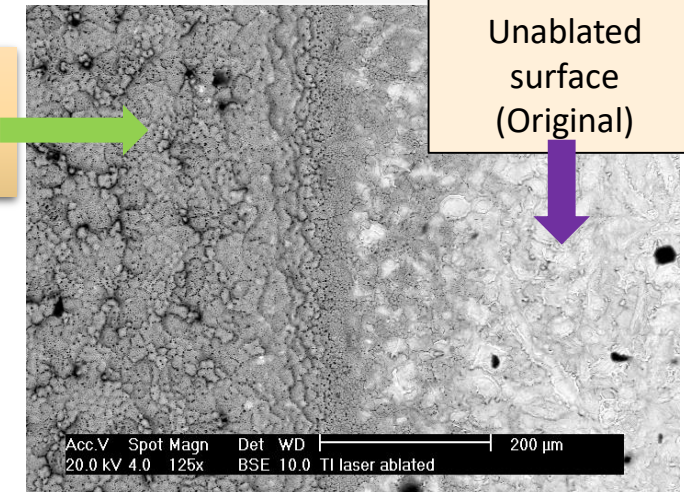
Amandeep Singh



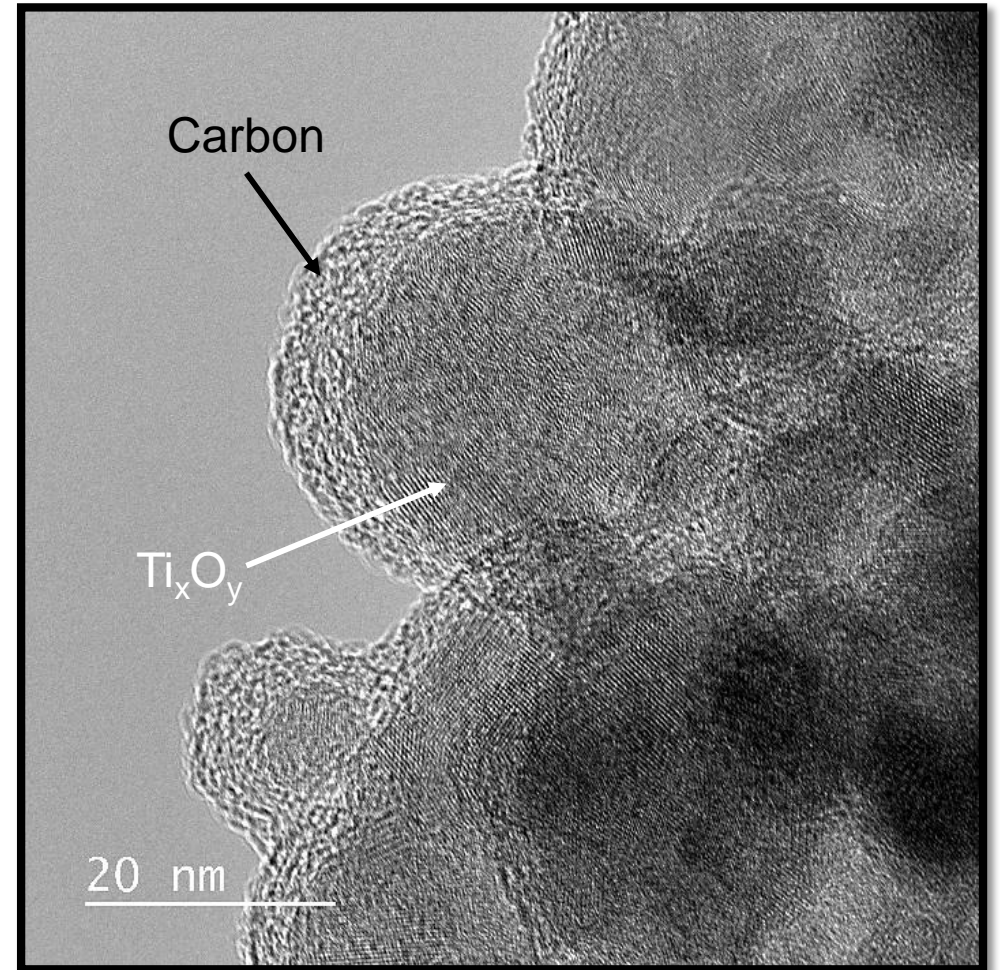
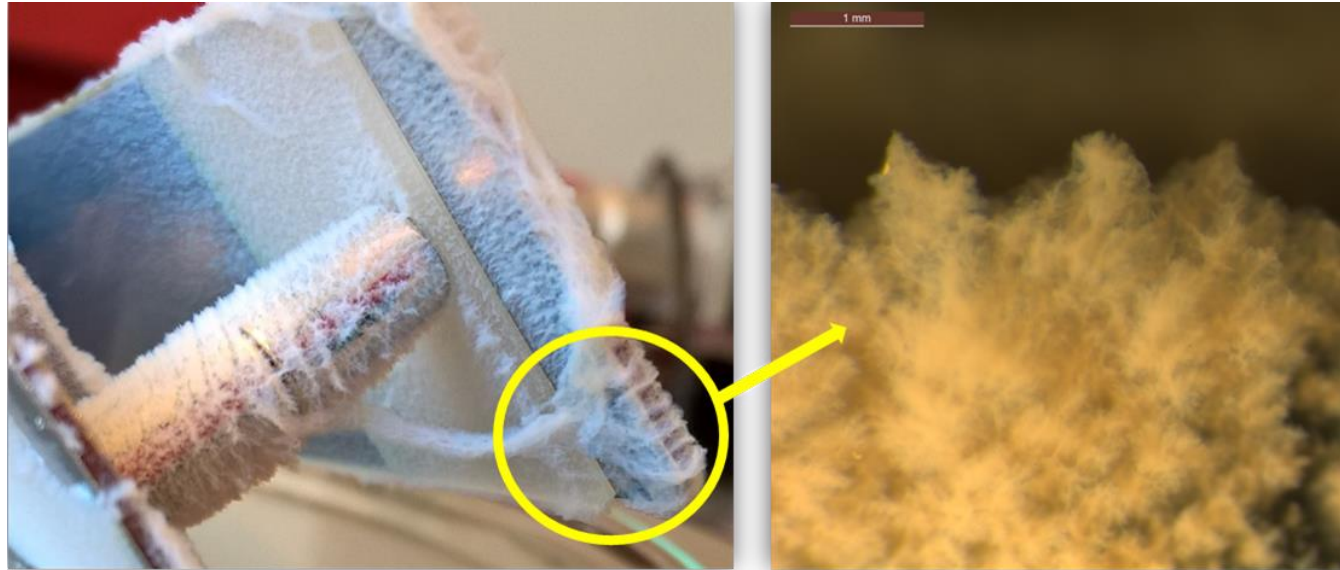
Ti<sub>x</sub>O<sub>y</sub>  
Nanoparticles



Ablated surface  
(Functionalized)

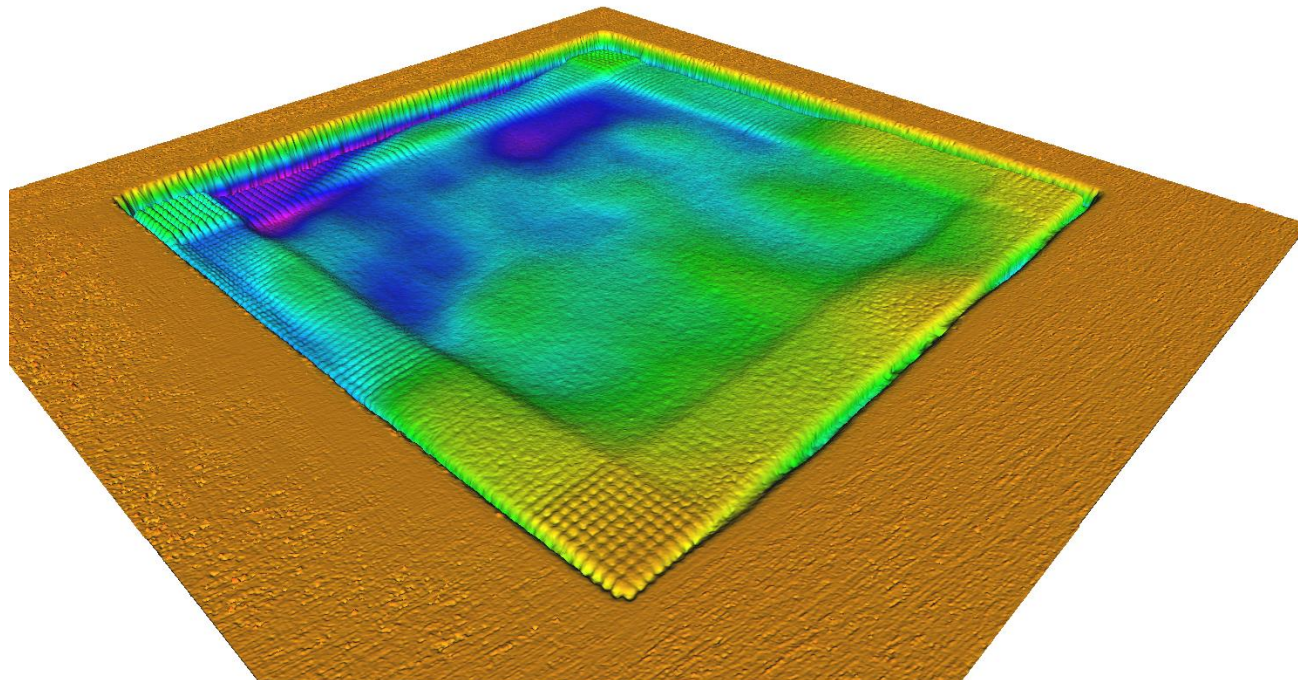


# $\text{Ti}_x\text{O}_y$ Nanoparticles





# Functionalized $\text{Ti}_x\text{O}_y$ surface

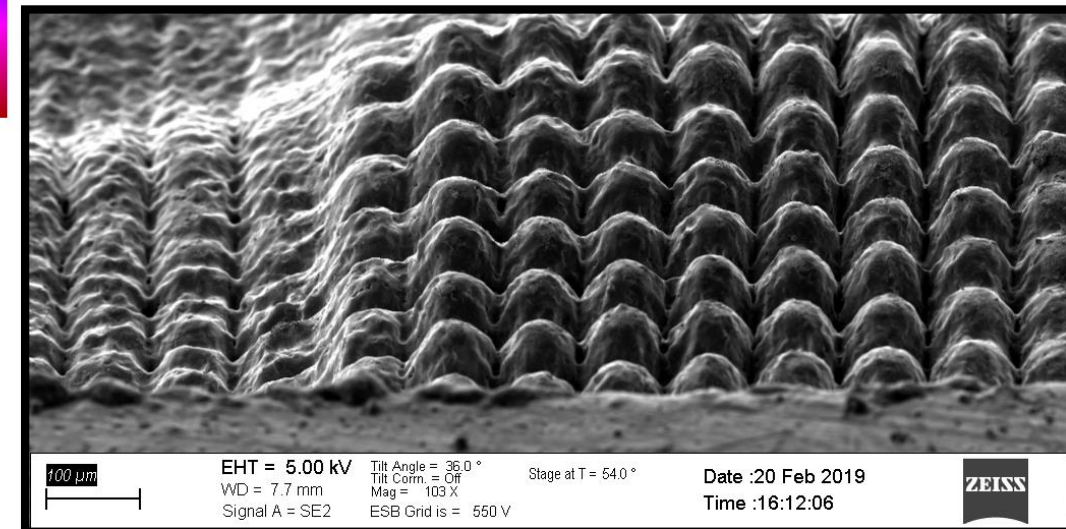
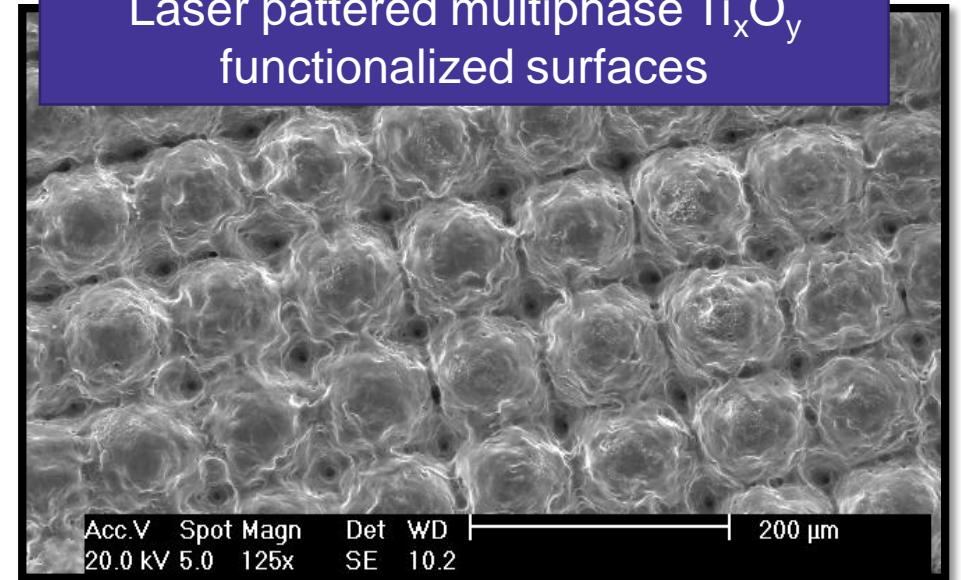


Possibility to make variable surface profiles by selecting laser parameters

Height  
 $\mu\text{m}$

0  
-50  
-100  
-150  
-200  
-250  
-300  
-350  
-400  
-450  
-500

Laser patterned multiphase  $\text{Ti}_x\text{O}_y$   
functionalized surfaces





# Conclusions

- Supercritical carbon dioxide is a versatile processing method with utilization potential in various industries
  - **Food**
  - **Pharmaceutical**
  - **Materials processing**
  - **Ceramic processing**
  - **Chemical engineering**
- Green processing method that utilizes already produced carbon dioxide (no extra emissions)
  - **In some applications reduces emissions**
- New application areas emerging constantly



# Thank you for your attention!



*Advanced ceramic materials group*



# Figure references:

CO2 Phase diagram: <https://www.nature.com/articles/35012181>

Caffeine extraction: doi:10.1111/j.1750-3841.2011.02304.x

Nanoparticles with scCO2: <https://www.eurekalert.org/features/doe/2004-08/dnnl-sfn082404.php>

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Painted galvanized sheet: <https://www.englertinc.com/blog/tag/metal-roof-colors/page/2/>